

A Novel Method for Constructions of β -D-Mannosidic, 2-Acetamido-2-deoxy- β -D-mannosidic, and 2-Deoxy- β -D-arabino-hexopyranosidic Units from the Bis(triflate) Derivative of β -D-Galactoside

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The useful constructions of β -D-mannosidic, 2-acetamido-2-deoxy- β -D-mannosidic, and 2-deoxy- β -D-arabino-hexopyranosidic units from the same intermediate, 2,4-bis(*O*-trifluoromethanesulfonyl) derivative of β -D-galactoside, were achieved in a stepwise inversion at C-4 and C-2 by using cesium acetate, Bu₄NBH₄ and Bu₄NN₃ in good yields. Convenient and practical protections of β -D-mannoside to the straightforward synthesis of antennary oligosaccharides also achieved by using cesium trifluoroacetate.

Despite of the recent explosive growth of oligosaccharide synthesis, the construction of β -D-mannosidic linkages remains a crucial step, far from being adequately solved in preparative terms. The various β -D-mannosyl donors available are accessible either by multistage synthesis only, or lack appreciable β -selectivity in glycosylations, or both.¹ Recent strategies for intramolecular aglycon delivery^{1,2} solve the β -selectivity problem, yet their practical utility for the synthesis of biologically relevant β -D-mannosides remains to be demonstrated. The applications of the different methodologies developed for C-2-epimerization of β -D-glucosides³ and for the β -D-mannosidase-promoted mannosyl transfer,⁴ which, although promising, has not attained the practically stage. The present most relevant method for the construction of β -D-mannosidic linkages appears to be an "indirect" one, involving β -D-glycosid-2-uloses as the key intermediates. These oxidation and reduction approaches have extensively used⁵⁻¹³ despite of that the stereoselectivity of the reduction is rarely very high. More recently, 3,4,6-tri-*O*-benzyl- α -D-arabino-hexopyranos-2-ulose bromide, a versatile glycosyl donor for efficient generation of β -D-mannosidic linkages, was reported¹⁴ as an excellent method.

In this paper, we would like to describe the efficient method for construction of β -D-mannosidic, 2-acetamido-2-deoxy- β -D-mannosidic, and 2-deoxy- β -D-arabino-hexopyranosidic units, those of which have been somewhat difficult to construct, in short steps and high yields from 3,6-di-*O*-pivaloyl-2,4-bis(*O*-trifluoromethanesulfonyl)- β -D-galactoside. The stepwise inversions of the bis(triflate) at C-4 and C-2 were achieved by the conditions employed. (Scheme 1) The selective protections of β -D-mannosidic unit for synthesizing high mannose sugar chain were also achieved by double inversion with cesium trifluoroacetate.

The key starting material, benzyl 3,6-di-*O*-pivaloyl- β -D-galactopyranoside (**2**) was prepared in the following way. Glycosidation of 1,2,3,4,6-penta-*O*-acetyl- β -D-galactopyranose

with benzyl alcohol, in the presence of trimethylsilyl triflate as promoter,¹⁵ gave benzyl 2,3,4,6-tetra-*O*-acetyl- β -D-galactoside in 80% yield. This was de-*O*-acylated with NaOMe in methanol (pH 9) to give the corresponding benzyl β -D-galactoside (**1**) in quantitative yield. Compound **1** was treated with bis(tributyltin) oxide¹⁶ (1.5 equiv.) under reflux in toluene, and then with pivaloyl chloride (3.0 equiv.) at r.t. in toluene to give the selectively protected derivative **2** in 85% yield. The pivaloyl group was used to distinguish it from acetyl groups. Compound **2** was treated with trifluoromethanesulfonic anhydride (3.0 equiv.) and pyridine in CH₂Cl₂ at 0 °C then at r.t. to give bis(triflate) **3** in 98% yield. In this work, **3** was prepared as a model compound, but naturally occurring compounds with other aglycons such as terpenes, steroids, and carbohydrates (especially blocked β -D-glucosaminide) may also be available as shown in Figure 1. Compound **3** was treated with CsOAc (1.5 equiv.) and 18-crown-6 in toluene at r.t. to give 4-*O*-monoacetyl derivative **4**, which is stable to purify on a column of silica gel, in 84% yield. Then, **4** was treated again with CsOAc at r.t. with ultrasonication (ca. 12 h) to give benzyl 2,4-di-*O*-acetyl-3,6-di-*O*-pivaloyl- β -D-mannopyranoside **5** in 94% yield, which was also obtained directly from **3** with 3 equiv. of CsOAc under the conditions with ultrasonication for 12 h in 93% yield. The above reaction carried out under reflux conditions (ca. 1 h) also gave **5** in 90% yield. In a similar way as mentioned above, **4** was treated with Bu₄NBH₄ or Bu₄NN₃ in benzene with ultrasonication to give the corresponding 2-deoxy derivative **6** in 82% yield or 2-azido-2-deoxy derivative **7** in 91% yield. Then, **7** was reduced in the presence of 5% Pd-C and H₂ in benzene (bubbling-through system) with stirring, followed by acetylation to give benzyl 2-acetamido-4-*O*-acetyl-2-deoxy-3,6-di-*O*-pivaloyl- β -D-mannopyranoside **8** in 88% yield. As mentioned above, the otherwise difficult constructions of β -mannosidic linkage of 1,2-*cis* relationship and 2-deoxy- β -D-mannosidic linkage were achieved easily via our indirect method involving stepwise nucleophilic substitution.

For synthesizing asparagine-linked sugar chains, proper protection of β -D-mannoside is required. Concerning this request, we examined the selective protection of benzyl β -D-mannoside by employing S_N2 inversion with cesium trifluoroacetate, because selective cleavage of acetyl and pivaloyl groups was difficult. The reaction of **3** with cesium trifluoroacetate and 18-crown-6 in toluene-DMF (3:1) at 80 °C gave a mixture of 2-*O*-, 4-*O*-, and 2,4-di-*O*-trifluoroacetyl derivatives. The mixed products were treated with aqueous sodium hydrogencarbonate in methanol gave benzyl 3,6-di-*O*-pivaloyl- β -D-mannoside **9** in 76% yield. Compound **9** was then treated with CH₂(OMe)₂ and P₂O₅ in (CH₂Cl)₂ to give the corresponding 2,4-bis(*O*-methoxymethyl) derivative in 93% yield. Deacylation of the above product with NaOMe in methanol gave benzyl 2,4-bis(*O*-methoxymethyl)- β -D-mannopyranoside **10** in quantitative yield. This methodology to the straightforward synthesis of antennary oligosaccharides, branched at the center β -D-mannosidic unit, seems to be useful for synthesizing important sugar units.

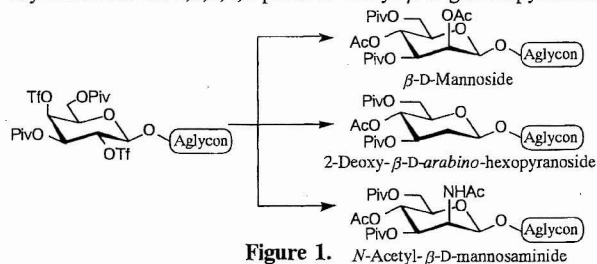
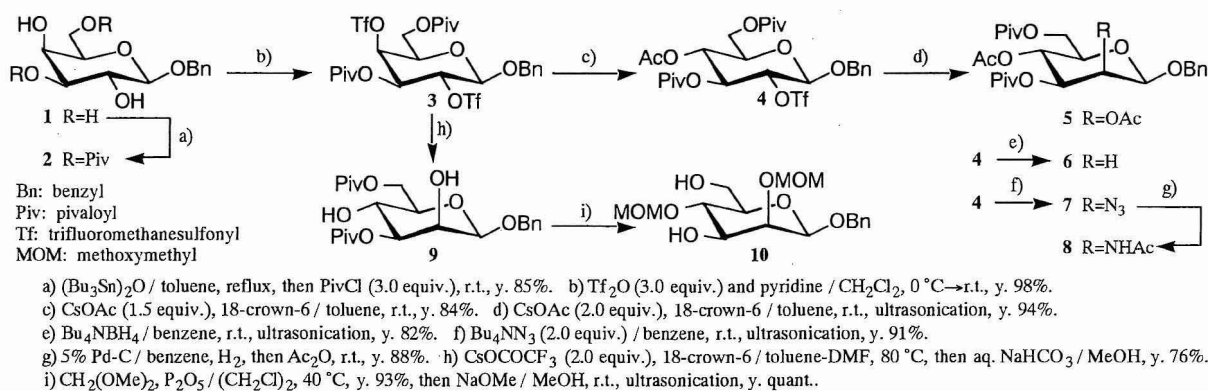


Figure 1.



Scheme 1.

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References and Notes

- For a recent review, see: E. Kaji and F. W. Lichtenthaler, *Trends Glycosci. Glycotechnol.*, **1993**, 121.
- F. Barresi and O. Hindsgaul, *J. Am. Chem. Soc.*, **113**, 9376 (1991); F. Barresi and O. Hindsgaul, *Synlett*, **1992**, 759; G. Stork and G. Kim, *J. Am. Chem. Soc.*, **114**, 1087 (1992).
- S. David, A. Malleron, and C. Dini, *Carbohydr. Res.*, **188**, 193 (1989); J. Alais and S. David, *Carbohydr. Res.*, **201**, 69 (1990); W. Günther and H. Kunz, *Carbohydr. Res.*, **228**, 217 (1992).
- N. Taubken, B. Sauerbrei, and J. Thiem, *J. Carbohydr. Chem.*, **12**, 651 (1993).
- G. Ekborg, B. Lindberg, and J. Lönnngren, *Acta Chem. Scand.*, **26** (8), 3287 (1972).
- N. K. Kochetkov, B. A. Dmitriev, N. N. Malysheva, A. Y. Chernyak, E. M. Klimov, N. E. Bayramova, and V. I. Torgov, *Carbohydr. Res.*, **45**, 283 (1975).
- M. A. E. Shaban and R. W. Jeanloz, *Carbohydr. Res.*, **52**, 103 (1976).
- M. A. E. Shaban and R. W. Jeanloz, *Carbohydr. Res.*, **52**, 115 (1976).
- C. D. Warren, C. Augé, M. L. Laver, S. Suzuki, D. Power, and R. W. Jeanloz, *Carbohydr. Res.*, **82**, 71 (1980).
- C. Augé, C. D. Warren, R. W. Jeanloz, M. Kiso, and L. Anderson, *Carbohydr. Res.*, **82**, 85 (1980).
- J. Kerékgyártó, J. P. Kamerling, J. B. Bouwstra, J. F. G. Vliegthart, and A. Lipták, *Carbohydr. Res.*, **186**, 51 (1989).
- J. Kerékgyártó, J. G. M. van der Ven, J. P. Kamerling, A. Lipták, and J. F. G. Vliegthart, *Carbohydr. Res.*, **238**, 135 (1993).
- K. K.-C. Liu and S. J. Danishefsky, *J. Org. Chem.*, **59**, 1892 (1994).
- F. W. Lichtenthaler and T. Schneider-Adams, *J. Org. Chem.*, **59**, (1994), in press; See also: F. W. Lichtenthaler, U. Kläres, M. Lergenmüller, and S. Schwidetzky, *Synthesis*, **1992**, 179; F. W. Lichtenthaler, E. Kaji, and S. Weprek, *J. Org. Chem.*, **50**, 3505 (1985); F. W. Lichtenthaler and E. Kaji, *Liebigs Ann. Chem.*, **1985**, 1659.
- H. Paulsen and M. Paal, *Carbohydr. Res.*, **135**, 53 (1984).
- T. Ogawa and M. Matsui, *Carbohydr. Res.*, **56**, C1 (1977).
- Physical data of each compound were as follows.
 2: mp $138-139^\circ\text{C}$; $^1\text{H NMR}$ $\delta=7.37-7.29$ (5H, m, Ph), 4.93 and 4.63 (1H x2, each d, $J_{A,B}=11.9\text{Hz}$, -CH₂-), 4.82 (1H, dd, $J_{3,2}=10.1\text{Hz}$, $J_{3,4}=3.4\text{Hz}$, H-3), 4.40 (1H, d, $J_{1,2}=7.6\text{Hz}$, H-1), 4.35 (1H, dd, $J_{6,5}=6.1\text{Hz}$, $J_{6,6}=11.6\text{Hz}$, H-6), 4.32 (1H, dd, $J_{6,5}=6.7\text{Hz}$, H-6), 3.96 (1H, m, H-4), 3.89 (1H, ddd, $J_{2,\text{OH}}=3.2\text{Hz}$, H-2), 3.74 (1H, m, H-5), 2.31 (1H, d, 2-OH), 2.18 (1H, d, $J_{\text{OH},4}=5.5\text{Hz}$, 4-OH), 1.25 and 1.22 (9H x2, each s, OPiv x2), 3: mp $69-73^\circ\text{C}$; $^1\text{H NMR}$ $\delta=7.41-7.30$ (5H, m, Ph), 5.29 (1H, dd, $J_{4,3}=2.7\text{Hz}$,

H-4), 5.13 (1H, dd, $J_{3,2}=10.5\text{Hz}$, H-3), 4.99 (1H, dd, $J_{2,1}=7.6\text{Hz}$, H-2), 4.91 and 4.69 (1H x2, each d, $J_{A,B}=11.5\text{Hz}$, -CH₂-), 4.69 (1H, d, H-1), 4.41 (1H, dd, $J_{6,5}=9.5\text{Hz}$, $J_{6,6}=14.2\text{Hz}$, H-6), 4.35-3.96 (2H, m, H-5 and H-6), 1.28 and 1.22 (9H x2, each s, OPiv x2), 4: mp $121-123^\circ\text{C}$; $^1\text{H NMR}$ $\delta=7.36-7.34$ (5H, m, Ph), 5.37 (1H, dd, $J_{4,3}=9.4\text{Hz}$, H-4), 5.09 (1H, dd, $J_{3,2}=9.8\text{Hz}$, H-3), 4.89 and 4.69 (1H x2, each d, $J_{A,B}=11.0\text{Hz}$, -CH₂-), 4.75 (1H, dd, $J_{2,1}=7.8\text{Hz}$, H-2), 4.64 (1H, d, H-1), 4.23 (1H, dd, $J_{6,5}=2.7\text{Hz}$, $J_{6,6}=12.5\text{Hz}$, H-6), 4.12 (1H, dd, $J_{6,5}=5.3\text{Hz}$, H-6), 3.70 (1H, ddd, H-5), 2.01 (3H, s, OAc), 1.25 and 1.18 (9H x2, each s, OPiv x2), 5: syrup; $^1\text{H NMR}$ $\delta=7.36-7.34$ (5H, m, Ph), 5.49 (1H, dd, $J_{2,1}=0.9\text{Hz}$, $J_{2,3}=3.4\text{Hz}$, H-2), 5.30 (1H, dd, $J_{4,3}=10.0\text{Hz}$, H-4), 4.96 (1H, dd, H-3), 4.88 and 4.65 (1H x2, each d, $J_{A,B}=12.3\text{Hz}$, -CH₂-), 4.62 (1H, d, H-1), 4.31 (1H, dd, $J_{6,5}=2.7\text{Hz}$, $J_{6,6}=12.2\text{Hz}$, H-6), 4.18 (1H, dd, $J_{6,5}=6.1\text{Hz}$, H-6), 3.46 (1H, ddd, H-5), 2.16 and 2.01 (3H x2, each s, OAc x2), 1.26 and 1.12 (9H x2, each s, OPiv x2), 6: mp $57-59^\circ\text{C}$ (not recrystallized); $^1\text{H NMR}$ $\delta=7.37-7.30$ (5H, m, Ph), 5.01 (1H, dd, $J_{4,3}=9.5\text{Hz}$, H-4), 4.94 (1H, ddd, $J_{3,2e}=5.2\text{Hz}$, $J_{3,2a}=11.5\text{Hz}$, H-3), 4.87 and 4.60 (1H x2, each d, $J_{A,B}=11.9\text{Hz}$, -CH₂-), 4.64 (1H, dd, $J_{1,2e}=2.0\text{Hz}$, $J_{1,2a}=9.6\text{Hz}$, H-1), 4.25 (1H, dd, $J_{6,5}=2.7\text{Hz}$, $J_{6,6}=12.1\text{Hz}$, H-6), 4.17 (1H, dd, $J_{6,5}=5.8\text{Hz}$, H-6), 3.62 (1H, ddd, H-5), 2.32 (1H, ddd, $J_{2e,2a}=12.5\text{Hz}$, H-2e), 2.01 (3H, s, OAc), 1.75 (1H, ddd, H-2a), 1.25 and 1.14 (9H x2, each s, OPiv x2), 7: syrup; $^1\text{H NMR}$ $\delta=7.37-7.35$ (5H, m, Ph), 5.29 (1H, dd, $J_{4,3}=9.8\text{Hz}$, H-4), 4.95 and 4.66 (1H x2, each d, $J_{A,B}=12.2\text{Hz}$, -CH₂-), 4.89 (1H, dd, $J_{3,2}=3.9\text{Hz}$, H-3), 4.64 (1H, d, $J_{1,2}=1.0\text{Hz}$, H-1), 4.27 (1H, dd, $J_{6,5}=2.4\text{Hz}$, $J_{6,6}=12.2\text{Hz}$, H-6), 4.12 (1H, dd, $J_{6,5}=5.9\text{Hz}$, H-6), 4.05 (1H, dd, H-2), 3.60 (1H, ddd, H-5), 2.01 (3H, s, OAc), 1.26 and 1.20 (9H x2, each s, OPiv x2), 8: mp $194-195^\circ\text{C}$; $^1\text{H NMR}$ $\delta=7.34-7.31$ (5H, m, Ph), 5.67 (1H, d, $J_{\text{NH},2}=8.8\text{Hz}$, NH), 5.29 (1H, dd, $J_{4,3}=9.8\text{Hz}$, H-4), 4.89 (1H, dd, $J_{3,2}=3.9\text{Hz}$, H-3), 4.84 and 4.61 (1H x2, each d, $J_{A,B}=12.3\text{Hz}$, -CH₂-), 4.76 (1H, ddd, $J_{2,1}=1.0\text{Hz}$, H-2), 4.64 (1H, d, H-1), 4.23 (1H x2, each d, $J_{6,5}=4.4\text{Hz}$, H-6 and H-6'), 3.60 (1H, ddd, H-5), 2.03 and 2.02 (3H x2, each s, OAc and NAc), 1.27 and 1.21 (9H x2, each s, OPiv x2), 9: mp $45-46^\circ\text{C}$ (not recrystallized); $^1\text{H NMR}$ $\delta=7.36-7.30$ (5H, m, Ph), 4.90 and 4.65 (1H x2, each d, $J_{A,B}=12.0\text{Hz}$, -CH₂-), 4.73 (1H, dd, $J_{3,2}=3.2\text{Hz}$, $J_{3,4}=9.8\text{Hz}$, H-3), 4.58 (1H, d, $J_{1,2}=0.9\text{Hz}$, H-1), 4.49 (1H, dd, $J_{6,5}=2.7\text{Hz}$, $J_{6,6}=12.0\text{Hz}$, H-6), 4.36 (1H, dd, $J_{6,5}=6.1\text{Hz}$, H-6'), 4.08 (1H, ddd, $J_{2,\text{OH}}=2.4\text{Hz}$, H-2), 3.92 (1H, ddd, $J_{4,5}=9.5\text{Hz}$, $J_{4,\text{OH}}=4.9\text{Hz}$, H-4), 3.94 (1H, ddd, H-5), 2.52 and 2.36 (1H x2, each d, OH x2), 1.25 and 1.24 (9H x2, each s, OPiv x2), 10: mp $119-120^\circ\text{C}$; $^1\text{H NMR}$ $\delta=7.37-7.29$ (5H, m, Ph), 4.93 and 4.65 (1H x2, each d, $J_{A,B}=12.2\text{Hz}$, -CH₂-), 4.90 and 4.83 (1H x2, each d, $J_{A,B}=6.7\text{Hz}$, -CH₂-), 4.83 and 4.71 (1H x2, each d, $J_{A,B}=6.7\text{Hz}$, -CH₂-), 4.56 (1H, d, $J_{1,2}=1.0\text{Hz}$, H-1), 4.02 (1H, dd, $J_{2,3}=3.4\text{Hz}$, H-2), 3.91 (1H, m, H-6), 3.88 (1H, d, $J_{\text{OH},3}=5.8\text{Hz}$, 3-OH), 3.82 (1H, m, H-6'), 3.68 (1H, dd, $J_{4,3}=9.5\text{Hz}$, H-4), 3.58 (1H, ddd, H-3), 3.46 and 3.44 (3H x2, each s, OMe x2), 3.30 (1H, ddd, $J_{5,6}=2.8\text{Hz}$, $J_{5,6}=5.4\text{Hz}$, H-5), 2.18 (1H, m, 6-OH).